REARRANGED LABDANE DITERPENOIDS FROM GALEOPSIS ANGUSTIFOLIA

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Abstract—From the aerial part of *Galeopsis angustifolia* collected in Spain, a labdane and three rearranged labdane diterpenoids have been isolated. The structures of these new substances have been established by chemical and spectroscopic means and by correlation with known compounds. A sample of the same species collected in Italy and other species of *Galeopsis* showed some remarkable chemical differences. The taxonomic significance of these results is discussed briefly.

INTRODUCTION

In a previous communication [1], we reported hispanolone (1) [2], galeopsin (2) and pregaleopsin as the major diterpenoid constituents of *Galeopsis angustifolia* growing in Spain. A study of the minor diterpene constituents of this plant has now allowed the isolation of four new diterpenoids, galeopsitrione (3, 15,16-epoxy-8,9-secolabdane-13(16),14-diene-7,8,9-trione), its hemiacetal derivative galeolone (4), galeopsinolone (5, 15,16-epoxy-6 β -hydroxy-8,9-seco-7,11-cyclolabdane-7(11),13(16),14-trien-9-one) and hispanone (6, 15,16-epoxylabdane-8,13(16),14-trien-7-one).

RESULTS AND DISCUSSION

The first of the new diterpenoids, galeopsitrione (3), C₂₀H₂₈O₄, had an IR spectrum which showed strong and broad ketone (1700 cm⁻¹) and furanoid (3150, 3130, 1503, 880 cm⁻¹) absorptions and no hydroxyl bands. The ¹H NMR spectrum showed signals for a β -substituted furan ring (two α -furan protons at δ 7.29 and 7.16, and a β furan proton at 6.21), three C-Me singlets (δ 1.22, 0.93 and 0.90) and a singlet of a methyl group adjacent to a carbonyl grouping (δ 2.35). Moreover, the ¹³C NMR spectrum showed signals corresponding to three ketone groups (δ 215.5, 199.0 and 197.1), and typical resonances for a β -substituted furan ring (Table 1). Two of the ketone groups form an α-diketone grouping, compound 3 showing UV absorption at $\lambda_{\rm max}$ 245 nm (log $\varepsilon=3.23$) which changed to $\lambda_{\rm max}$ 265 (log $\varepsilon=3.64$) and 300 nm (log $\varepsilon=3.34$) in basic media [3], and it formed a quinoxaline derivative when treated with σ -phenylendiamine. In the ¹H NMR spectrum of this quinoxaline derivative there was a paramagnetic shift ($\Delta\delta + 0.45$) of the signal assigned to the methyl group adjacent to a ketone, thus, a -CO-CO-Me structural moiety was established for galeopsitrione.

Treatment of galeopsitrione (3) with aqueous hydrochloric acid in ethanol yielded a unique product,

 $\rm C_{20}H_{24}O_2$, for which structure 7 was assigned on the basis of its spectroscopic properties (see Experimental). The formation of 7 from galeopsitrione (3) may be rationalized in terms of an aldol condensation between C-7 and C-11 followed by an alkylation reaction between C-16 and C-8.

All the above deductions on the structures of galeopsitrione (3) and its derivative (7), as well as the absolute configuration of these compounds, were confirmed as follows. When galeopsin (2) [1] was heated at 190° for 10 min under N_2 and without solvent, a retroaldol

reaction occurred to give 8 [1], which on reaction with t-BuOK in benzene quantitatively yielded 9 [4, 5]. Oxidation of this compound with chromium trioxide in pyridine gave 10, which, on treatment with hydrochloric acid in ethanol, was quantitatively transformed into a substance identical in all respects (mp, mmp, $[\alpha]_D$, IR, UV, 1 H NMR and mass spectra) with compound 7. Moreover, no transformation was observed when the hydroxyketone, 9, was treated under the same conditions as the diketone, 10.

Table 1. ¹³C NMR data for compounds 3, 5 and 12 (25.2 MHz, CDCl₃, TMS as int. standard)

Carbon No.	3	5	12
1	36.5 t*	35.9 t	33.5 t
2	18.1 t	18.5 t	18.3 t
3	40.9 t	44.2 t	41.6 t
4	34.2 s	34.6 s	33.6 s
5	44.2 d	51.2 d	48.3 d
6	37.8 t*	67.0 d	27.7 t*
7	199.0 s‡	156.8 s	158.0 s
8	197.1 s†	24.7 t	28.0 t*
9	215.5 s	203.8 s	204.1 s
10	51.6 s	44 .6 s	44 .1 s
11	33.8 t	131.5 s	130.5 s
12	19.2 t	20.8 t	20.7 t
13	124.0 s	123.8 s	123.7 s
14	110.9 d	111.2 d	111.0 d
15	142.5 d	142.5 d	142.1 d
16	138.9 d	139.3 d	138.8 d
17	23.6 q	24.5 q	17.1 q
18	33.0 q	32.1 q	32.2 q
19	22.6 q	22.2 q	22.1 q
20	17.0 q	12.5 q	12.3 q

^{*, †}These assignments may be interchanged.

The C-7-C-9 hemiacetal of the C-7 enolic form of galeopsitrione (4, galeolone) was also present. It showed IR absorptions for hydroxyl (3420 cm⁻¹), furanoid (3150, 1505, 880 cm⁻¹) and α,β -unsaturated ketone (1670, 1650 cm⁻¹) groups. Its ¹H NMR spectrum showed signals (see Experimental) for a β -substituted furan ring, three C-Me groups, a methyl group adjacent to an α,β -unsaturated ketone and an AX system attributed to the C-6 and C-5 protons (δ 5.58, d, and 3.12, d, respectively, $J_{5,6} = 11$ Hz). Moreover, UV absorptions at λ_{max} 211 nm (furan) and λ_{max} 265 nm (α -ether- α,β -unsaturated ketone) supported structure 4 for this substance. Treatment of galeolone (4) with hydrochloric acid in ethanol also yielded compound 7.

Another diterpenoid isolated from G. angustifolia, galeopsinolone (5), was a $C_{20}H_{28}O_3$ compound. Its IR spectrum showed hydroxyl (3470 cm⁻¹), furanoid (3150, 3130, 1505, 880 cm⁻¹) and α,β -unsaturated ketone (1650, 1635 cm⁻¹) absorptions. The presence of an α,β -unsaturated ketone in galeopsinolone was also confirmed by its UV absorption at $\lambda_{\rm max}$ 238 nm (log $\varepsilon=3.74$). The ¹H NMR spectrum of this substance showed signals in complete agreement with structure 5: a β -substituted furan ring, three C-Me singlets (C-18-C-20), an ethyl side chain attached to a fully substituted sp^2 carbon atom (δ 1.12, 3H, t, t = 7.5 Hz, 3H-17, and 2.43, 2H, t, t

= 7.5 Hz, 2H-8), an AB system at δ 3.55 and 3.29 (J = 15 Hz) due to the C-12 methylene group, and a one-proton doublet at δ 4.58, which must be assigned to the equatorial geminal proton of a C-6 β (axial) hydroxyl group, because it is only coupled (J = 3.5 Hz) with the C-5 α (axial) proton (δ 1.52, d, J = 3.5 Hz).

Chromium trioxide-pyridine oxidation of galeopsinolone (5) yielded the ene-dione derivative 11 (λ_{max} 253.5 nm, $\log \varepsilon = 3.98$), the ¹H NMR spectrum of which showed a singlet at δ 2.50 for its C-5 α proton. Furthermore, comparison of the ¹³C NMR spectra (Table 1) of galeopsinolone (5) and compound 12, a substance previously obtained from hispanolone (1) [4, 5], clearly confirmed structure 5 for this new diterpenoid. In particular, the β -effect on C-5 ($\Delta\delta$ + 2.9), the δ -effects on C-1, C-3 and C-17 ($\Delta\delta$ + 2.4, + 2.6 and + 7.4, respectively) and the γ -effects on C-4 and C-8 ($\Delta\delta$ + 1.0 and - 3.3, respectively) rigorously established the existence of a C-6 β (axial) hydroxyl group in the molecule of galeopsinolone (5).

The last diterpenoid isolated from *G. angustifolia* has been named hispanone and it was identical in all respects with a compound previously obtained by dehydration of hispanolone (1) [4, 5]. Thus, hispanone possesses structure 6.

From a biogenetic point of view, it is important to note that the hydrocarbon skeleton of galeopsitrione (3), galeolone (4) and galeopsinolone (5) must be derived from the labdane skeleton of galeopsin (2). A retroaldol reaction produces the 8,9-secolabdanes 3 and 4, and a subsequent regioselective aldol condensation generates the rearranged labdane 5.

On completion of this part of the work we had the opportunity of examining some samples of G. angustifolia collected in northern Italy. Extraction of the material gave a diterpene fraction containing pregaleopsin [1] and galeolone (4), but not the other diterpenoids found in the Spanish samples (compounds 1-3, 5 and 6). On the contrary, we isolated from the Italian samples the triterpene hederagenin $(3\beta,23$ -dihydroxy-olean-12-en-28-oic acid) [6] and the flavones salvigenin (5-hydroxy-6,7,4'trimethoxyflavone) and galangustin (5,7-dihydroxy-8,4'dimethoxyflavone) [7], i.e. compounds that are absent from the Spanish samples. Therefore, some differences in chemical contents seem to occur between G. angustifolia growing in Spain and northern Italy; they could be attributed to varietal differences or to diversities of climate, humidity, soil, etc.

These results led us to carry out a chemotaxonomic survey of other representatives of the genus Galeopsis. We examined G. ladanum L., G. speciosa Miller, G. terahit L. subsp. silvestris Schlecht and G. tetrahit subsp. reichenbachii Reuter. None of these plants contained diterpenes, triterpenes or flavonoids. This negative result, however, may be of some chemotaxonomic significance. The genus Galeopsis [8, 9] is divided into two subgenera: Ladanum (G. segetum = G. dubia = G. ochroleuca, G. pyrenaica, G.ladanum = G. intermedia, G. angustifolia and G. reuteri) and Galeopsis (G. speciosa, G. pubescens, G. tetrahit and G. bifida). Only G. ochroleuca has been previously studied [10] and reported to contain some flavonoids. The occurrence of diterpenoids only in G. angustifolia could suggest the need to relocate this species in a separate subgenus.

Further chemotaxonomic investigation on other species of *Galeopsis* should be useful, also taking into account their possible pharmacological value as evidenced

by the current use of some species in traditional medicines [11].

EXPERIMENTAL

Mps are uncorr. Elemental analyses were carried out in Madrid with the help of an automatic analyser. Assignments of ¹³C NMR shifts were made with the aid of off-resonance and noise-decoupled ¹³C NMR spectra. Plant materials of G. angustifolia and G. ladanum were collected in June 1981 near Molina de Aragón (Guadalajara) and Rodiezmo (León), respectively, both in Spain, and voucher specimens were deposited in the Herbarium of the Faculty of Pharmacy (Madrid, Computense University, Spain). Plant materials of G. speciosa were collected in June 1981 in Austria, and of G. angustifolia, G. tetrahit subsp. silvestris and subsp. reichenbachii in June 1981 in northern Italy, and voucher specimens were deposited in the Herbarium of the Istituto di Botanica, University of Milano, Italy.

The previously known compounds were identified by their physical (mp, $[\alpha]_D$) and spectroscopic (IR, ¹H NMR, MS) data and by comparison with authentic samples.

Compounds 1, 2, 6, 8 and 12 have been previously described [1, 2, 4, 5].

Isolation of the diterpenoids. Dried and finely powdered aerial parts (2.1 kg) of G. angustifolia, collected in Spain, were extracted (×3) with Me₂CO (10 l.) at room temp. for 1 week. After filtration the extracts were evaporated to dryness under red. pres. and low temp. (26°). The residue (87 g) was chromatographed on a Si gel (Merck, No. 7734, deactivated with 15% H₂O) column (1.5 kg). Elution with n-hexane and n-hexane-EtOAc (19:1 and 9:1) gave, in order of elution, galeopsitrione (3, 240 mg), hispanone (6, 150 mg), galeolone (4, 36 mg), hispanolone (1, 4.3 g) [1, 2], galeopsinolone (5, 180 mg), galeopsin (2.2 l g) [1] and pregaleopsin (2.5 g) [1].

Identical treatment of a sample of G. angustifolia collected in Italy yielded galeolone (4), pregaleopsin [1], hederagenin [6], salvigenin and galangustin [7]. Samples of G. speciosa, G. ladanum, G. tetrahit subsp. silvestris and G. tetrahit subsp. reichenbachii were also investigated, but none of these plants contained triterpenes or diterpenes or flavonoids in detectable amounts.

Galeopsitrione (3). A syrup, $[\alpha]_D^{25}$ 1.5038; $[\alpha]_D^{24} - 7.5^{\circ}$ (CHCl₃; c 0.61); IR $v_{\text{max}}^{\text{NaCl}}$ cm⁻¹: 3150, 3130, 1700 (br), 1503, 1465, 1385, 1355, 1170, 1070, 1027, 880, 790, 730; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm ($\log \varepsilon$): 221.5 (3.70), 245 (3.23); $\lambda_{\text{max}}^{\text{MeOH}} + \text{NaOMe}$ nm ($\log \varepsilon$): 220 (3.74), 265 (3.64), 300 (3.34); ¹H NMR (90 MHz, CDCl₃): see Results and Discussion; ¹³C NMR (25.2 MHz, CDCl₃): see Table 1; EIMS (direct inlet) 75 eV, m/z (rel. int.): 332 [M] ⁺ (6), 317 (3), 289 (100), 271 (7), 261 (4), 247 (6), 229 (8), 195 (9), 177 (8), 161 (7), 149 (8), 123 (60), 109 (24), 95 (33), 81 (84), 69 (36), 55 (33), 43 (60), 41 (51). (Found: C, 72.03; H, 8.53. $C_{20}H_{28}O_4$ requires: C, 72.26; H, 8.49 %)

Galeolone (4). A syrup, $[\alpha]_{20}^{20} + 31.4^{\circ}$ (CHCl₃; c 0.59); IR $v_{\rm nac}^{\rm Max}$ cm⁻¹: 3420, 3150, 2930, 2880, 1670, 1650, 1505, 1450, 1390, 1350, 1240, 1025, 880, 785; UV $\lambda_{\rm max}^{\rm EtOH}$ nm (log ε): 211 (3.79), 265 (4.01); ¹H NMR (90 MHz, CDCl₃): δ 7.27 (1H, m, $W_{4} = 4.5$ Hz, H-15 or H-16), 7.17 (1H, m, $W_{4} = 4.5$ Hz, H-16, or H-15), 6.22 (1H, m, $W_{4} = 4$ Hz, H-14), 5.58 (1H, d, J = 11 Hz, H-6), 3.12 (1H, d, J = 11 Hz, H-5), 2.73 (4H, complex signal, 2H-11 and 2H-12), 2.37 (3H, s, 3H-17), C-Me singlets at 1.10 (3H), 0.93 (3H) and 0.87 (3H); EIMS (direct inlet) 75 eV, m/z (rel. int.): 332 [M]⁺ (31), 317 (2), 314 (1), 299 (5), 289 (19), 271 (3), 209 (27), 191 (14), 163 (12), 153 (15), 149 (15), 127 (34), 123 (52), 113 (30), 109 (36), 95 (70), 81 (82), 69 (56), 55 (37), 53 (35), 43 (100). (Found: C, 72.08; H, 8.43, C₂₀H₂₈O₄ requires: C, 72.26; H, 8.49 %)

Galeopsinolone (5). Mp 153-154° (from EtOAc-n-hexane);

 $[\alpha]_D^{21} - 117.9^{\circ}$ (CHCl₃; c 0.273); IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3470, 3150, 3130, 2990, 2960, 2930, 2900, 2870, 1650, 1635, 1505, 1455, 1375, 1290, 1210, 1165, 1040, 1025, 995, 950, 880, 770; UV λ_{max}^{MeOH} nm (log ϵ): 217 (3.72), 238 (3.74); ¹H NMR (90 MHz, CDCl₃): δ 7.25 (1H, m, $W_{\frac{1}{2}} = 4.5 \text{ Hz}$, H-15 or H-16), 7.13 (1H, m, $W_{\frac{1}{2}} = 4.5 \text{ Hz}$, H-16 or H-15), 6.20 (1H, m, $W_1 = 4.5$ Hz, H-14), 4.58 (1H, d, J = 3.5 Hz, equatorial H-6), 3.55 and 3.29 (an AB system, J = 15 Hz, 2H-12), 2.43 (2H, br q, J = 7.5 Hz, 2H-8), 1.52 (1H, d, J = 3.5 Hz, H-5), 1.12 (3H, t, J = 7.5 Hz, 3H-17), C-Me singlets at 1.38 (3H), 1.35 (3H) and 1.10 (3H). Some of these assignments (H-5, H-6, 2H-8, 2H-12 and 3H-17) were confirmed by double resonance expts. ¹³C NMR (25.2 MHz, CDCl₃): see Table 1; EIMS (direct inlet) 75 eV, m/z (rel. int.): 316 [M]⁺ (90), 301 (16), 298 (2), 287 (28), 283 (10), 269 (12), 255 (12), 231 (17), 219 (15), 203 (16), 192 (25), 164 (32), 109 (32), 91 (43), 81 (100), 69 (50), 55 (80), 53 (60), 43 (75). (Found: C, 75.69; H, 8.96. C₂₀H₂₈O₃ requires: C, 75.91; H, 8.92 %.)

Hispanone (6). Mp $58-60^{\circ}$ (from MeOH); $[\alpha]_D^{18} + 38.7^{\circ}$ (CHCl₃; c 1.15) (lit. [4, 5]: mp $58-60^{\circ}$, $[\alpha]_D + 39.7^{\circ}$). The IR, UV, ¹H NMR, ¹³C NMR and MS of the natural diterpenoid (6) were identical with the previously reported data for the synthetic compound [4, 5].

Quinoxaline derivative of galeopsitrione. To a soln of 3 (20 mg) in $H_2O-HOAc$ (9:1, 3 ml) a crystal (ca 20 mg) of σ -phenylendiamine was added. The soln was then heated at 100° for 10 min. Work-up in the usual manner yielded the quinoxaline derivative (23 mg, after two crystallizations from MeOH), mp $101-103^{\circ}$; IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3150, 3080, 2940, 2880, 1695, 1575, 1505, 1495, 1445, 1397, 1390, 1030, 880, 770, 730; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 237.5 (4.23), 308 sh (3.66), 319 (3.75), 330 sh (3.61); ¹H NMR (90 MHz, CDCl₃): δ 8.10-7.45 (4H, complex signal, aromatic protons), 7.17 (1H, m, $W_{\downarrow} = 5$ Hz, H-15 or H-16), 6.80 (1H, m, $W_{\downarrow} = 4$ Hz, H-16 or H-15), 5.90 (1H, m, $W_4 = 4$ Hz, H-14), 2.80 (3H, s, 3H-17), C-Me singlets at 1.53 (3H), 1.07 (3H) and 1.03 (3H); EIMS (direct inlet) 75 eV, m/z (rel. int.): 404 [M] (55), 389 (15), 281 (100), 254 (8), 252 (10), 171 (30), 158 (85), 123 (45), 102 (10), 95 (15), 81 (70), 69 (20), 41 (50). (Found: C, 77.26; H, 7.83; N, 7.08. $C_{26}H_{32}O_2N_2$ requires: C, 77.19; H, 7.97; N, 6.93%.

Compound 7 from galeopsitrione (3) and galeolone (4). To a soln of 3 (50 mg) in EtOH (10 ml) cone. HCl (1 ml) was added, and the soln heated under reflux for 5 hr. The soln was then cooled, diluted with H₂O and extracted with CHCl₃. Evaporation of the solvent yielded pure 7 (42 mg), mp 107–108° (from MeOH); $[\alpha]_D^{20}$ -4.4° (CHCl₃; c 0.86); IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3160, 3130, 3015, 3005, 2980, 2960, 2920, 2880, 2850, 1680, 1620, 1590, 1540, 1460, 1390, 1380, 1335, 1305, 1200, 1135, 1085, 1035, 910, 880, 790, 775, 740, 700; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 236.5 (4.52), 258 (3.76), 272 (3.62), 310 (3.06); ¹H NMR $(90 \text{ MHz}, \text{CDCl}_3)$: $\delta 8.20 (1\text{H}, s, \text{H-12}), 7.60 (1\text{H}, s, \text{H-12})$ d, J = 2.4 Hz, H-15), 6.79 (1H, d, J = 2.4 Hz, H-14), 3.05 (1H, dd,part A of an ABX system, $J_{AB} = 17 \text{ Hz}$, $J_{AX} = 5 \text{ Hz}$, H-6 α), 2.83 $(1H, dd, part B of an ABX system, J_{BX} = 11 Hz, H-6\beta), 2.49 (3H,$ s, 3H-17), 2.07 (1H, dd, part X of an ABX system, H-5), C-Me singlets at 1.13 (6H) and 1.06 (3H); EIMS (direct inlet), 75 eV, m/z(rel. int.): 296 [M] + (100), 281 (55), 255 (45), 213 (36), 211 (35), 186 (36), 144 (18), 128 (15), 115 (20), 55 (22), 41 (40). (Found: C, 80.86; H, 8.26. C₂₀H₂₄O₂ requires: C, 81.04; H, 8.16%)

Identical treatment of galeolone (4) also yielded 7.

Compounds 8 and 9 from galeopsin (2). The preparation of 8 from galeopsin (2) has been previously described [1]. Treatment of 8 with t-BuOK in C_6H_6 soln, at room temp. for 12 hr quantitatively yielded 9 as an inseparable mixture of the C-8 epimers [1]. Quantitative transformation of galeopsin (2) into 9 was achieved by treatment of 2 with 5% ethanolic KOH for 24 hr at room temp. Compound 9 was a syrup, IR $\nu_{\rm max}^{\rm NaCl}$ cm⁻¹: 3440, 3150, 2920, 2880, 2860, 1665, 1635, 1503, 1455, 1370, 1150, 1060, 1025, 880, 775; UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ϵ): 218 (3.81), 242 (3.95); $^{\rm 1}$ H

NMR (90 MHz, CDCl₃): δ 7.25 (1H, m, $W_{\frac{1}{2}} = 4.5$ Hz, H-15 or H-16), 7.10 (1H, m, $W_{\frac{1}{2}} = 4.5$ Hz, H-16 or H-15), 6.17 (1H, m, $W_{\frac{1}{2}} = 4.5$ Hz, H-14), 4.93 (1H, q, J = 6 Hz, H-8), 3.51 and 3.19 (an AB system, J = 15 Hz, 2H-12), 1.29 (3H, d, J = 6 Hz, 3H-17), C-Me singlets at 1.05 (3H), 1.01 (3H) and 0.97 (3H); EIMS (direct inlet), 75 eV, m/z (rel. int.): 316 [M] + (70), 301 (8), 298 (22), 283 (20), 273 (32), 146 (28), 109 (34), 95 (22), 91 (40), 81 (100), 69 (70), 55 (54).

After treatment of 9 with HCl-EtOH as previously described for galeopsitrione (3), only the starting material was recovered.

Ene-dione 10. CrO₃-pyridine oxidation of 9 (300 mg) yielded 10 (290 mg) as a syrup, $[n]_{\rm b}^{17}$ 1.5317; $[\alpha]_{\rm b}^{21}$ - 72.9° (CHCl₃; c 0.635); IR $v_{\rm max}^{\rm NaCl}$ cm⁻¹: 3160, 3150, 2940, 2880, 2860, 1695, 1680, 1505, 1460, 1430, 1360, 1230, 1170, 1030, 880, 780; UV $\lambda_{\rm max}^{\rm MeOH}$ nm (log ε): 214.5 (3.92), 242 (3.88); ¹H NMR (90 MHz, CDCl₃): δ 7.21 (1H, m, $W_{\frac{1}{2}}$ = 4.5 Hz, H-15 or H-16), 7.15 (1H, m, $W_{\frac{1}{2}}$ = 4.5 Hz, H-16 or H-15), 6.20 (1H, m, $W_{\frac{1}{2}}$ = 4.5 Hz, H-14), 3.43 and 3.27 (an AB system, J = 14 Hz, 2H-12), 2.31 (3H, s, 3H-17), C-Me singlets at 1.03 (3H), 1.00 (3H) and 0.93 (3H); EIMS (direct inlet), 75 eV, m/z (rel. int.): 314 [M]⁺ (66), 299 (22), 285 (7), 271 (5), 147 (12), 109 (10), 91 (16), 81 (28), 69 (18), 55 (23), 43 (100), 41 (40). (Found: C, 76.23; H, 8.49. C₂₀H₂₀O₃ requires: C, 76.40; H, 8.34%)

Compound 7 from 10. Treatment of 10 with HCl-EtOH as previously described for galeopsitrione, quantitatively yielded 7 (identical mp, mmp, TLC, $[\alpha]_D$, IR, UV, ¹H NMR and MS).

CrO₃-pyridine oxidation of galeopsinolone (5) to give 11. This was performed in the usual manner. Compound 11 is a syrup, $[n]_D^{15}$ 1.5266; $[\alpha]_D^{21} - 82.3^{\circ}$ (CHCl₃; c 0.35); IR $v_{\text{max}}^{\text{NaCl}}$ cm⁻¹: 3150, 2940, 2880, 1690, 1670, 1620, 1505, 1460, 1380, 1240, 1185, 1175, 1025, 995, 880, 780; UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ε): 215.5 (3.77), 253.5 (3.98); ¹H NMR (90 MHz, CDCl₃): δ 7.27 (1H, m, $W_4 = 4.5$ Hz, H-15 or H-16), 7.17 (1H, m, $W_4 = 4.5$ Hz, H-16 or H-15), 6.18 (1H, m, $W_4 = 4.5$ Hz, H-14), 3.62 and 3.42 (an AB system, J = 14 Hz, 2H-12), 2.50 (1H, s, H-5), 2.42 (2H, g, g = 7.5 Hz, 2H-8), 1.00 (3H, g t, g = 7.5 Hz, 3H-17). C-Me singlets at 1.28 (3H) and 1.13 (6H); EIMS (direct inlet), 75 eV, m/z (rel. int.): 314 [M]⁺ (100), 299 (16), 285 (8), 281 (9), 271 (12), 231 (14), 229 (12), 162 (10), 109 (27), 95 (20), 91 (30), 81 (68), 67 (31), 55 (56), 53 (54), 43 (36), 41 (92).

(Found: C, 76.23; H, 8.29. $C_{20}H_{26}O_3$ requires: C, 76.40; H, 8.34 $^{\circ}_{(2)}$)

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